VINCARUBINE, A NOVEL BISINDOLE ALKALOID FROM VINCA MINOR L.

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The structure of a novel bisindole alkaloid vincarubine (1) was adduced from the spectral data. Vincarubine is the first dimeric alkaloid isolated from the leaves of lesser periwinkle.

Vincarubine, the dark-red indole alkaloid was separated by counter-current distribution followed by column chromatography of the mixture of tertiary strong bases fraction from the leaves of lesser periwinkle (Vinca minor L.). This cryptocrystalline base, $C_{A3}H_{50}N_AO_6$ (718.9), m.p. 170°C (decomp.), $[d]_D^{E3}$ -550 \pm 10 (ethanol) displayed absorption bands in the UV region at \bigwedge nm (log \mathcal{E} , m². mol⁻¹) 273 (3.21), 340 (3.29) and 480 (2.40). Its IR spectrum (KBr, \widetilde{v} , cm⁻¹) indicated the presence of an ester group (1730), an d, β -unsaturated carbonyl grouping (1680) and a double bond (1605). The ^{'H} NMR spectrum showed signals attributable to four isolated protons of an aromatic system, three methoxyl, one $N-CH_3$, one ethylidene, and one ethyl groupings. Upon $NABH_A$ reduction in methanol vincarubine (1) formed a yellow compound, which, in turn, underwent a spontaneous oxidation in air to afford the starting material. Therefore, 1 was reduced with sodium dithionate and the product was directly acetylated to the stable 2, $C_{45}H_{54}N_{4}O_{7}$ (762.9). The UV spectrum of 2 revealed absorption bands at λ_{max} , nm (log ε , m². $mol⁻¹$) 259 (3.11) and 347 (3.46). Addition of methanolic HCl to the sample was associated with both hypsochromic shift to 343 nm and an appearance of a shoulder at 320 nm. This phenomenon is reported^{1,2} as diagnostic of the vincorane type of alkaloid having a $C_KH_S-N-C-N$ grouping. This assignment was also backed by the shift of the 13 C NMR signal of the carbon between two nitrogen atoms from δ 104.0 ppm for 1 to 94.7 ppm for 2, the latter value being close to that published for vincorine³ and $(-)$ -1-norvincorine (3). A contrast of the 13 C NMR spectra of 1 and 2 with those of flexicorine and dihydroflexicorine⁵ showed the vincorine moiety of both types of alkaloids to be closely related. The absence of $N_{(1^*)}$ -CH₃ in the structures of 1 and 2 caused, as follows also from the analysis of the spectra of

vincorine and (-)-1-norvincorine⁴ that carbons $C_{(3')}$ resonate at d 27.8 and 26.4 ppm, respectively and the respective carbons $C_{(6)}$ at d 41.0 and 41.1 ppm. Assignment of all signals of 1 was confirmed by analysis of homoand heterocorrelated 2D spectra. As a result, the signals ascribed to $\mathfrak{C}_{\left(\begin{smallmatrix}3&\bullet\end{smallmatrix}\right)}$ and $C_{\ell, \zeta}$, for flexicorine' should be interchanged.

Carbon	1^{a}	$2^{\mathbf{a}}$	4°	Carbon	1^a	$2^{\mathbf{a}}$	$\boldsymbol{\hat{z}}^{\text{c}}$
S.	166.6	167.3	167.8	2^{\prime}	104.0	94.7	94.4
3	50.3	50.4	51.7	3 ²	27.8	26.4^{+}	26.6
5	51.8	51.7	50.7	5 ²	53.8	54.5	56.0
6	47.0	47.2	45.3	6 ²	41.0	41.1	41.3
$\overline{7}$	56.0	56.2	55.5	$7\,$	56.8	57.7	57.8
8	130.6	130.3	138.0	\mathbf{g}^{\star}	158.0	136.4	139.4
9	122.8	123.2	120.5	9 ²	123.4	118.8	112.2^+
10	116.7	118.9	120.5	10 ²	187.0	140.8	153.1
11	157.7	156.8	127.4	11 ²	144.7	130.9	111.9^{+}
12	93.5	93.3	109.3	12 ²	131.1	111.7	109.3
13	148.9	147.3	143.4	13 ²	164.5	145.4	141.8
14	22.0	22.2	22.2	14 ²	26.7	26.5^{+}	26.6
15	32.6	32.6	32.9	15 [′]	35.7	35.1	35.3
16	94.2	93.5	92.8	16 ²	50.3	50.6	50.6
17	30.2	30.0	25.6	17 ²	172.6	173.3	173.5
18	7.1	7.2	7.3	18 [′]	13.8	13.6	13.5
19	29.4	29.3	29.3	19 ²	123.1	123.3	122.0
20	37.5	37.6	38.2	20 ²	139.2	138.2	138.2
21	75.0	75.2	72.7	21 ²	59.2	57.7	58.1
$N-CH_3$	35.9	36.1		OCH ₃			54.6
OCH ₃	56.1	56.1			$COOCH_{3}$ 51.9	51.7	51.7
COOCH ₃	168.0	168.0	169.2	$2H_3$ COO		20.9	
COOCH ₃	50.9	50.9	50.9	CH ₃ COO		169.7	

TABLE 1. ¹³C MMR chemical shifts (δ , ppm) of compounds 1 - 4

^ain **CDCl at 303** K, internal standard tetramethylsilane; ^b **4 3 ref.";** $^{\circ}$ ref.⁴; $^+$ may be interchanged.

The mass spectrum of 1 disclosed peaks associated with M+ at m/z **718** (42%) and further ones corresponding to $M + 2$ (100%), $M + 14$ and $M + 2 + 14$ ions. The $M + 2$ ion formed by a thermal in situ hydrogenation of the quinonimine moiety of the **molecule** was observed in the spectrum of flexicorine, as well. The $M + 14$ peak reflected the thermal transmethylation of bisindole alkaloids having methoxycarbonyl and amine substitutions as e.g. voacamine $^\circ$ or vinblastine. Other prominent peaks appeared at $\frac{m}{2}$ M + 2 - 15 (3%),

 $M - 15$ (4%), $M + 2 - 28$ (4%), $M + 2 - \text{COOCH}_3$ (8%), $M - \text{COOCH}_3$ (12%), 258 (2%), 124 (42%), 107 (45%) and 92 (63%). The dihydroacetate $\frac{2}{2}$ revealed peaks at m/z 776 (M + 14, 2%), 762 (M⁺, 21%), 720 (M - C₂H₂0, 2%), 381 (the vincorane moiety of the molecule) and 124 (100%). The 1 $\frac{1}{4}$ intensity ratio of the peak at $\underline{m}/\underline{z}$ 762 to that at 124 indicated the vincadifformine type of alkaloids⁸. The IR absorption band at 1680 cm^{-1} of compound <u>2</u> remained unchanged, this being in favour of an d, β -unsaturated carbonyl group in this part of the alkaloid vincarubine. This band is diagnostic of anilinoacrylate alkaloids, as evidenoed by the **'II-** and '3C XMR spectra analysis. The selective transfer of polarization' through H₍₉₎ at *d* 7.0 ppm afforded signals at 157.7 (C₍₁₁₎), 148.9 (C₍₁₃₎) and 144.7 ppm. The latter is due

to $C_{(11')}$ of the vincorane moiety, and therefore, an opposite arrangement of $C_{(8')}$ and $C_{(11')}$ signals was suggested in contrast to⁵. The remaining signals of the alicyclic carbons are in line, as far as the quantity and multiplicity are concerned, with those¹⁰ of vincadifformine (4); some differences in signal positions might be due to different substitution at $N_{(1)}$. The band at 2776 cm⁻¹ in the IR spectra of 1 and 2 comparable with that at 2760 cm^{-1} of compound 4 favourizes the trans orientation¹¹ of the $H_{(21)}$ with respect to the lone electron pair at $N_{(4)}$.

The presented data entitle us to ascribe structure 1 to vincarubine, which is the first bisindole alkaloid isolated from the leaves of Vinca minor L.

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